



Elements of Chemical and Physical Geology: V.1

Gustav Bischof

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This historic book may have numerous typos, missing text or index. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. 1854. Not illustrated. Excerpt: ... I. By the separation of carbonic acid and carburetted hydrogen. ii. By the separation of carbonic acid from the elements of wood, and by the oxidation of hydrogen by external oxygen. iii. By the separation of carbonic acid and water from among the elements of wood. These three cases belong to the definite problems, for which formulae may be constructed. The case in which carbonic acid, carburetted hydrogen, and water are separated, is an indefinite problem which admits of several solutions. Among the possible modes in which the change may take place, there is none in which a larger quantity of carbon is separated from the wood, and lost, than in the first of the above cases; and, on the other hand, none in which less is lost than in the third case. This loss of carbon approaches nearer to the maximum the greater the quantity of carburetted hydrogen mixed with the disengaged water and carbonic acid, and nearer to the minimum in the opposite case. If the quantities of the elements are represented--In wood. In coal, or any product of the alteration of wood. Carbon = a. = a Hydrogen = b. = /3 Oxygen = c. = « and if in the first-mentioned case x = the oxygen, and y = the hydrogen which must be separated, we have then the following proportions:--a--0-375 x--Zy: b--y: c--x = a: f3: «. Hence it follows that, («-313) c-(a-3b) a x = a-0-375 s-3/3 x--c) (i + 8 b y = 8 If, in the second case, x = the oxygen which is disengaged in combination with the carbon, and y = the hydrogen oxidized by external oxygen, we obtain the proportions, a--0-375 x: b--y: c--x = a:j3:8. Hence it follows that, ac--Ha = a-0-375 8 (x--c) j8 + 8 4 y = 1 If, in the third case, x = the oxygen disengaged in combination with carbon, and y = the hydrogen disengaged in combination with ...

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